

Exponential Decay

We use the same exponential model for exponential decay as we did for exponential growth.

$$\frac{dy}{dt} = ky \Rightarrow y = Ce^{kt}$$

C = initial amount

k = rate of decay

The only difference is that, in exponential growth, $k > 0$, and, in exponential decay, $k < 0$.

Example 1: The population P of a species of bird is decreasing at a rate that is proportional to the population itself. If $P = 5000$ when $t = 3$ and $P = 4000$ when $t = 4$, what is the population when $t = 9$?

$$P = Ce^{kt}$$

$$5000 = Ce^{3k} \Rightarrow 5000e^{-3k} = C$$

$$4000 = Ce^{4k} \Rightarrow 4000 = 5000e^{-3k}e^{4k}$$

$$\Rightarrow \frac{4}{5} = e^k \Rightarrow k = \ln\left(\frac{4}{5}\right)$$

$$\Rightarrow 5000 = Ce^{3\ln(4/5)}$$

$$\Rightarrow C = 5000e^{-3\ln(4/5)}$$

$$P(t) = 5000e^{-3\ln(4/5)}e^{\ln(4/5)t}$$

$$P(9) \approx \boxed{1311 \text{ birds}}$$

Half-Life Formula

The half-life of a substance is the amount of time it takes for half of the initial amount to decay. The following relationship holds between the half-life and the rate of decay, k .

$$k = \frac{\ln\left(\frac{1}{2}\right)}{\text{half-life}}$$

Example 2: The radioactive isotope ^{239}Pu has a half-life of approximately 24,100 years. After 2,000 years there are 5 grams of ^{239}Pu left.

(a) What was the initial quantity?

$$\text{Half-life} \Rightarrow k = \frac{\ln(1/2)}{24100}$$

$$P = ce^{kt}$$

$$5 = ce^{\frac{\ln(1/2)}{24100}(2000)}$$

$$\Rightarrow c = 5e^{-\frac{\ln(1/2)}{24100}(2000)} \approx \boxed{5.296 \text{ g}}$$

(b) How much remains after 5,000 years? $-\ln(1/2) = \ln(2)$

$$P = ce^{kt} \Rightarrow P_{(t)} = 5e^{\ln(2)\left(\frac{2000}{24100}\right)} e^{\frac{\ln(1/2)}{24100}t}$$

$$P(5000) \approx \boxed{4.587 \text{ g}}$$

DIY

1. The radioactive isotope ^{14}C has a half-life of approximately 5,715 years. A piece of charcoal contains only 25% as much of the radioactive carbon as a piece of modern charcoal. How old is this sample of charcoal?

$$\text{Half-life} \Rightarrow k = \frac{\ln(1/2)}{5715}$$

$$\begin{aligned} &\hookrightarrow 25\% \text{ of initial amount} \\ &= 25\% \text{ of } C = 0.25C. \end{aligned}$$

$$0.25C = Ce^{\frac{\ln(1/2)}{5715}t} \Rightarrow 0.25 = e^{\frac{\ln(1/2)}{5715}t}$$

$$\Rightarrow \ln(0.25) = \frac{\ln(1/2)}{5715}t \Rightarrow \frac{5715 \ln(0.25)}{\ln(1/2)} = t$$

$$\Rightarrow \boxed{t = 11,430 \text{ years}}$$